

Lipase biodiversity

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Abstract

Industries prefer biocatalysts rather than chemical catalyst. Lipase a biocatalyst is a versatile enzyme that not only hydrolyzes the esters of long chain aliphatic acids form glycerol at oil or water interface but also involved in hydrolysis, transesterification, alcoholysis, and aminolysis. Lipases are widely distributed in microorganisms, plants and animals. Among them microbial lipases are preferred because of easily obtainable. Lipases are used in many fields like food, dairy, detergent, pharmaceutical, agrochemical and oleochemical industries. Based on the data compiled it reveals that the contribution of bacterial lipases is 45%, fungal 21%, animal 18%, plants 11% and algae 3%. This article provides information about comparative account of bacterial, fungal, plant and animal origin lipases along with their biochemical profiles. It also focuses on the need in search of algal lipases.

Keywords: Enzyme, lipase, microorganisms, plants and animals

Introduction

Lipases are the special kind of esterases belong to subclass 1 of hydrolytic enzyme class 3 and have been assigned sub-sub class 3.1.1 due to their specificity for carboxylic acid ester bonds. Naturally, lipase acts on glycerides as they possess a chiral alcohol moiety. They are also useful for the resolution or asymmetrization of ester bearing a chiral alcohol moiety (Aravindan, 2007). The biological function of lipase is to catalyze the hydrolysis of triacylglycerols to give free fatty acid, diacylglycerols, mono-acylglycerols and glycerol. The lipases available from various sources have considerable variation in their reaction specificities. Some lipases have affinity for short chain fatty acid (acetic, butyric, capric acid or decanoic acid etc.) while others have preference for unsaturated fatty acids (oleic, linolenic acid etc.) and many other are non specific and randomly split the fatty acids from the triglycerides. Lipase possesses the unique features of interface between an aqueous and non aqueous phase, provide a new understanding of a rapidly moving field. In the present article a data on lipase of bacterial, fungal, plant and animal origin has been compiled. However, more emphasis has been made towards algal origin which is ephemeral.

Sources of lipase

Lipase occurs widely in nature; however microbial lipases are commercially significant because of low production cost, greater

stability and wider availability than other sources. Few review articles were published (Pahoja & Sethar 2002; Gupta & Rathi 2004; Aravindan *et al.*, 2007) on lipase sources along with its industrial applications. Fig.1 illustrates the biodiversity of lipases with biological origin and Fig.2 shows the number of characterized and uncharacterized lipases. The characterization is based on pH, temperature and molecular weight of the enzyme.

Bacterial lipases

Many bacterial lipases are well studied compared to plants and animals. Bacterial lipase is a glycoprotein but some extracellular bacterial lipases are lipoprotein. The organisms are normally grown on nutrient medium containing carbon (oil, sugar and mixed carbon sources), nitrogen, phosphorus sources and mineral salts whereas the production of lipases mostly depends on inducer such as triglycerides, bile salts and glycerol. Lipases from *Pseudomonas* were probably the first studied and have preponderant role in industries, later on *Achromobacter sp.*, *Alcaligones sp.*, *Pseudomonas sp.*, *Staphylococcus sp.*, and *Chromobacterium sp.*, have been exploited for production of lipases. Lang *et al.* (1996) derived crystal structure of the lipases form *Chromobacterium viscosum*. The sources and properties of bacterial lipases are given in Table 1(a). These lipases are characterized for pH, temperature, Pi and molecular weight from both gram positive and gram negative bacteria. It is evident from the Table 1(a) that, pH range is between 4.0 to 10.0; temperature range is between 27 to 80°C;

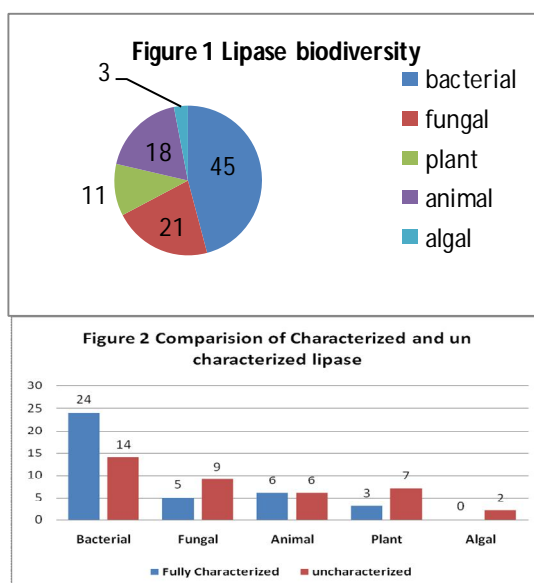




Table 1(a). Biochemical profile of bacterial lipases

Source	pH	Temp.	MW	Reference
<i>Acinetobacter calcoaceticus</i>	8.0	35	30.5	Brune & Gotz (1992)
<i>A. calcoaceticus</i>	5.0	47	23	Dharmsthiti <i>et al.</i> (1998)
<i>Acinetobacter sp.</i> CR9	8.0	40	--	Kasana <i>et al.</i> (2008)
<i>Acinetobacter sp.</i> RAG-1	9.0	62	33	Snellman <i>et al.</i> (2002)
<i>Alcaligenes sp.</i>	9.0	55	--	Brune & Gotz (1992)
<i>Bacillus alcalophilus</i>	10	60	--	Ghanem <i>et al.</i> (2000)
<i>B. coagulans</i> BTS-3	8.5	55	31	Kumar <i>et al.</i> (2005)
<i>B. licheniformis</i> strain H1	10	55	--	Khyami-Horani (1996)
<i>B. megaterium</i> AKG-1	7.0	55	--	Sekhon <i>et al.</i> (2004)
<i>B. pumilus</i> B26	8.5	35	--	Kim <i>et al.</i> (2002)
<i>B. stearothermophilus</i>	7.4	68	43	Kim <i>et al.</i> (2000)
<i>B. subtilis</i> 168	9.9	35	19	Lesuisse <i>et al.</i> (1993)
<i>B. thermocatenulatus</i>	8.5	70	--	Schmidt <i>et al.</i> (1996)
<i>B. thermoleovorans</i> CCR11	9.5	60	11	Lelie <i>et al.</i> (2005)
<i>B. thermoleovorans</i> ID-1	7.5	75	34	Lee <i>et al.</i> (1999)
<i>Bacillus sp.</i> LBN 4	7.0	50	--	Bora & Kalita, (2009)
<i>Bacillus sp.</i>	5.6	30	22	Sugihara <i>et al.</i> (1991)
<i>Bacillus sp.</i> strain 398	8.2	65	50	Kim <i>et al.</i> (1994)
<i>Bacillus sp.</i> strain A30-1	7.2	60	65	Wang <i>et al.</i> (1995)
<i>Bacillus sp.</i> THLO 27	7.0	70	69	Dharmsthiti <i>et al.</i> (1999)
<i>Burkholderia cepacia</i>	7.5	38.5	--	Gupta <i>et al.</i> (2001)
<i>Pseudomonas aeruginosa</i>	9.3	50	30	Lidija <i>et al.</i> (2009)
<i>P. aeruginosa</i> EF2	9.0	50	29	Gilbert <i>et al.</i> (1991)
<i>P. cepacia</i>	5.0	60	--	Dunhaupt <i>et al.</i> (1991)
<i>P. fluorescens</i>	9.0	52.5	33	Kojima <i>et al.</i> (1994)
<i>P. fluorescens</i> MC50	8.5	35	55	Brune & Gotz (1992)
<i>P. fragi</i>	8.2	65	55	Brune & Gotz (1992)
<i>Pseudomonas sp.</i> KWI-56	6.2	52.5	33	Brune & Gotz (1992)
<i>Pseudomonas sp.</i>	8.0	52.5	30	Dong <i>et al.</i> (1999)
<i>Pseudomonas sp.</i> strain KB 700A	8.2	35	--	Rashid <i>et al.</i> (2001)
<i>Pyrococcus furiosus</i>	7.0	30	28	Chandrayan <i>et al.</i> (2008)
<i>Rhodotorula pilimanae</i> lipase I	4.0	47.5	176	Muderhwa <i>et al.</i> (1986)
<i>Rhodotorula pilimanae</i> lipaseII	7.0	47.5	21.4	Muderhwa <i>et al.</i> (1986)
<i>Salinivibrio sp.</i> strain SA-2	7.7	50	--	Mohammad <i>et al.</i> (2008)
<i>Staphylococcus epidermidis</i>	6.0	40	--	Joseph <i>et al.</i> (2006)
<i>S. haemolyticus</i>	8.5	28	45	Byung <i>et al.</i> (2000)
<i>Thermobifida fusca</i>	8.0	60	29	Chen <i>et al.</i> (2008)
Thermophilic <i>Bacillus sp.</i> J33.	--	60	45	Nawani & Kaur (2004)

Temp. = Temperature in °C, MW = Molecular weight in kDa.

whereas, molecular weight varies from 11 to 176 kDa.

Fungal lipases

Fungal lipases have benefits over bacterial lipases due to their low cost of extraction, thermal and pH stability, substrate specificity and activity in organic solvents. Lipase producers are widespread in the fungal kingdom. The chief producers of lipases are *Aspergillus sp.*, *Candida sp.*, *Mucor sp.*, *Rhizopus sp.*, have been studied in great details. The thermophilic *Mucor pusillus* is well known as a producer of thermostable extracellular lipase and from *M. miehei* two isoenzymes with slightly different isoelectric points could be isolated. The sources and biochemical properties of fungal lipases are given in Table 1(b). Fungal lipases are characterized for pH, temperature and molecular weight. It is interesting to note that, pH range is between 4.0 to 11.0, temperature range is between 25 to 60°C whereas, molecular weight varies from 27 to 120 kDa.

Plant lipases

In plants mostly lipases are present in the form of food reserve tissues of growing seedlings or especially in those which contains large amount of triacylglycerols. Lipase activity in plant seeds increases during germination because the triacylglycerols are converted to soluble sugars by the action of lipase which is then transported to the growing tissues to supply structural carbon and energy to provide support for the growth of young plants. The sources and properties of plant lipases are given in Table 1(c). Plant lipases are characterized for pH, temperature and molecular weight. It is interesting to note that, pH range is in between 4.0 to 8.0, temperature range is in between 25 to 60°C, whereas, molecular weight varies from 40 to 143 kDa. This data indicates that relatively plant lipases are slightly different from bacterial and fungal lipases.

Animal lipases

Animals are also rich sources of lipases but due to the availability of microbial lipases they are rarely studied, but still they have been isolated from many insects, fishes, mammals. Animal lipase plays an important role in digestion of lipids in biological system (Walton & Cowey 1984). Fats required special digestive action before absorption because the end products must be carried in water medium (blood and lymph) in which fats are not soluble. Although little actual fat digestion occurs in the stomach, gastric lipase does digest already emulsified fats such as in egg yolk and cream. The detail biochemical profile of animal lipase is given in Table 1(d).

Properties of lipases

Types of reaction catalyzed by lipase

Lipases are stable and rugged enzyme that act on lipids as well on wide variety of natural and artificial reactant since it has ability to catalyze diversified reaction, few of them are explained below:

Acidolysis: It is the process of reacting an acid with an ester. Acidolysis between Triolein and short chain fatty acid by lipase is carried in organic solvents (Tsuzuki, 2005), *Aspergillus oryzae*, *Rhizomuco miehei* and *Candida cylindracea* lipases were used. In his experiments he observed ten kinds of lipases as biocatalysts for the incorporation of short chain fatty acids

Table 1(b). Biochemical profile of fungal lipases

Source	pH	Temp.	MW	Reference
<i>Aspergillus carneus</i>	9.0	37	27	Saxena <i>et al.</i> (2003)
<i>A. niger</i>	5.2	47.5	--	Namboodiri <i>et al.</i> (2000)
<i>A. niger</i> NCIM1207	8.5	50	--	Mhetras <i>et al.</i> (2008)
<i>Aureobasidium pullulans</i>	7.8	35	--	Kudanga <i>et al.</i> (2006)
<i>Candida cylindracea</i>	7.2	45	120	Ghosh <i>et al.</i> (1996)
<i>C. rugosa</i>	7.0	30	117	Pernas <i>et al.</i> (2001)
<i>Fusarium solani</i>	7.25	25	--	Poulsen <i>et al.</i> (2005)
<i>Neurospora crassa</i>	7.0	30	54	Kundu <i>et al.</i> (1987)
<i>Penicillium nitroaeducens</i>	11.0	33.5	--	Ghosh <i>et al.</i> (1996)
<i>Pichia burtonii</i>	6.5	45	51	Sharma <i>et al.</i> (2001)
<i>Rhizomucor miehei</i>	8.0	40	--	Herrgard <i>et al.</i> (2000)
<i>Rhizopus homothallicus</i>	--	50	29.5	Diaza <i>et al.</i> (2006)
<i>R. oryzae</i>	7.5	35	32	Sharma <i>et al.</i> (2001)
<i>Thermoactinomyces sp.</i>	8.0	60	--	Khudary <i>et al.</i> (2003)

Temp. = Temperature in °C, MW = Molecular weight in kDa

Table 1(c). Biochemical profile of plant lipases

Source	pH	Temp.	MW	Reference
<i>Caesalpinia bonducella</i> seeds	6.8	60	--	Pahoja <i>et al.</i> (2001)
<i>Carissa carandas</i> fruit	5.0	30	--	Mala <i>et al.</i> (1995)
Castor bean	9.0	25	62	Maeshima <i>et al.</i> (1985)
<i>Cucumis melo</i>	5.0	40	--	Akhtar <i>et al.</i> (1978)
<i>Hibiscus cannabinus</i>	5.0	40	--	Akhtar <i>et al.</i> (1979)
<i>Lycopersicon esculentum</i>	8	25	--	Matsui <i>et al.</i> (2004)
<i>Moringa olifera</i>	5.0	40	--	Dahot <i>et al.</i> (1987)
Rice bran	7.8	38.5	40	Aizono <i>et al.</i> (1973)
<i>Triticum aestivum</i>	8.0	37	143	Kapranichikov <i>et al.</i> (2004)
<i>Lycopersicon esculentum</i>	8	25	--	Matsui <i>et al.</i> (2004)

Temp. = Temperature in °C, MW = Molecular weight in kDa.

Table 1(d). Biochemical profile of animal lipases

Source	pH	Temp.	MW	Reference
Chicken Adipose	5.2	--	86	Marit <i>et al.</i> (1997)
<i>Cyprinion macrostomus</i>	7.5	37	51	Deuerlu <i>et al.</i> (2002)
<i>Homo sapiens</i>	7.5	37	---	Jocken <i>et al.</i> (2008)
<i>Homo sapiens gastric</i>	4.5	37	10	Carriere <i>et al.</i> (2000)
<i>Homo sapiens pancreatic</i>	7.0	37	50	Carriere <i>et al.</i> (2000)
<i>Oncorhynchus mykiss</i> trout	7.0	15	42	Harmon <i>et al.</i> (1991)
Rat adipose tissues	--	--	85	Belfrage <i>et al.</i> (1977)
<i>Rattus norvegicus</i>	8.6	--	180	Jensen <i>et al.</i> (1981)
Rohu (<i>Labea rohita</i>)	7.0	37	--	Nayak <i>et al.</i> (2004)
<i>Scorpio maurus</i>	9.0	37	50	Zouari <i>et al.</i> (2005)
<i>Sus scrofa</i>	8.0	40	240	David <i>et al.</i> (1998)
trout adipose	--	--	48	Sheridan <i>et al.</i> (1989)

Temp. = Temperature in °C, MW = Molecular weight in kDa.

(acetic, propionic and butyric acids) into triolein in order to produce one kind of reduced calorie structured lipids.

Transesterification: It is hydrolysis of triglycerides in the presence of alcohol to form methylester and glycerol. A new enzymatic route for methylesters production from soybean oil was suggested by Xu *et al.* (2003) and found that Novozyme 435 (immobilized *Candida antarctica* lipase) gave the highest methyl ester yield of 92% under optimum conditions and transesterification of 30%, based on oil weight molar ratio of methylacetate and oil of 12:1, temperature of 40°C and reaction time of 10 hours.

Esterification: A reaction of an alcohol with an acid to produce an ester and water is called as esterification. Unal (1998) investigated the effect of molecular sieves on

the esterification reaction between lauric acid and geraniol in isooctane catalyzed by immobilized lipase.

Intertransesterification: It is one of the most important processes for modifying the physicochemical characteristic of oils and fats. During transesterification, fatty acids are exchanged within and among triacylglycerols until thermodynamic equilibrium are reached. Most vegetable oils are unspecific in chemical composition. To widen their use, vegetable oils are modified chemically by interesterification. Usmani *et al.* (2010) worked on lipase catalysed interesterification for the production of oleochemicals from non-traditional oils.

Aminolysis

Aminolysis is the conversion of amines and alcohols into amides and esters or it is any chemical reaction in which a molecule is split into two parts by reacting with a molecule of ammonia or an amine. Couturies (2009) described the lipase catalyze chemoselective aminolysis of various aminoalcohols with fatty acids, they used *Candida antarctica* lipase and developed a solvent free enzymatic process for the production of fatty alcohol amides. The aminolysis of linoleylethyl esters with several aminoalcohols from C2 to C6 ethanolamine, aminopropanol, 3-amino-1,2-propanediol, 2-amino-1, 3 propanediol, 4-amino-1-butanol etc. are widely used aminoalcohols.

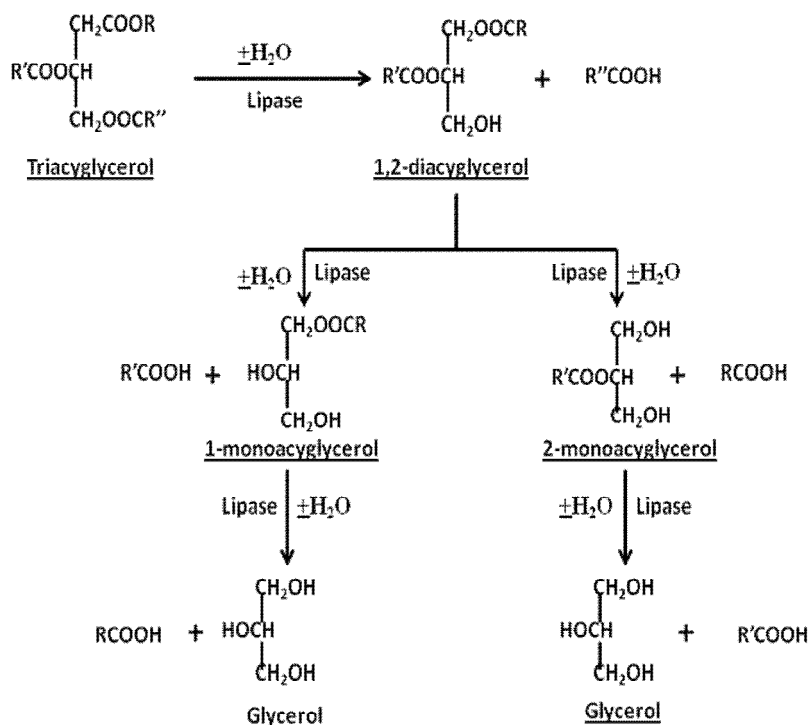
Lipase mediated conversion of natural oils: Amro *et al.* (2009) suggested that *Pseudomonas aeruginosa* has the ability to degrade castor oil. Rao (2008) worked on production of biodiesel using vegetable oil and ethyl acetate as acyl acceptor.

Hydrolysis: Triglycerides are composed of three glycerol and three fatty acids. Lipase hydrolysis or degrades triglycerides into its component parts of fatty acids and glycerol. Hills and Beevers (1987) suggested that

lipase hydrolyzes triacylglycerol to glycerol (Fig.3) and fatty acids which are converted to sugars and support growth of young plants.

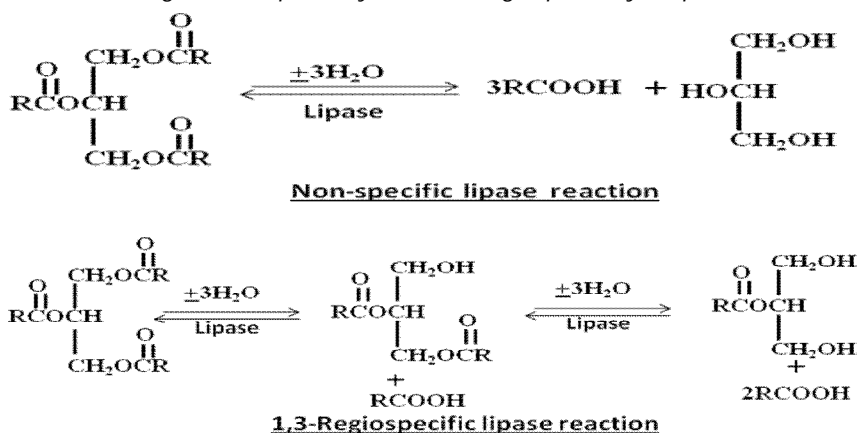
Alcoholysis: Hadzir *et al.* (2001) studied the reaction between Triolein and oleyl alcohol catalyzed by Lipozyme and Novozyme to produce wax esters. The best conditions tested to produce wax ester were incubation time of 5 hours, temperature of 50°C for Lipozyme and 60°C for Novozyme. Ghosh *et al.* (1996) investigated lipase catalyzed alcoholysis of soy phospholipids to simultaneously make lysophospholipids and fatty acid esters. Alcoholysis was carried out by stirring a mixture of soy phospholipids fatty acids and alcohols in equimolar

Fig.3. Reaction mechanism of lipase



proportions with esters and 10% of *Mucor miehei* lipase, in the presence of hexane as solvent at 55°C for 24 hours.

Fig. 4. Non-specificity and 1, 3-Regiospecificity of lipase



Substrate specificity of lipases

Lipases are often crucial to their application in industries and laboratories. Different lipases appear too specific in splitting various fatty acids. According to Tsujisaka, et al., (1977) the specificity of lipase is controlled by the molecular properties of enzyme, structure of substrate and factors affecting binding of enzyme to the substrate. Lipase act on substrate in specific and non-specific manner (Fig.4), resulting in complete or hydrolysis of triglycerides into free fatty acids and glycerol or along with triglycerides, monoacylglycerides and diacylglycerides, fatty acids and glycerol's are also formed (Aravindan *et al.*, 2007).

Specificity is shown in both the way with respect to either fatty acyl or alcohol part of their substrates (Macrae & Hammond, 1985). Lipase also showed both regiospecificity and stereospecificity with respect to alcohol moiety of their substrate (Ghosh, 1989). Lipase can be divided into two groups on basis of the regiospecificity exhibited on acyl glycerol substrate (Macrae & Hammond, 1985). First group lipase catalyses the complete breakdown of triacylglycerol to glycerol and free fatty acids together with diacylglycerols and monoacylglycerols as intermediates in the reaction, example of such lipase are isolated from *Candida cylindracea* (Kugumiya, 1986). Second type of lipase release fatty acids regiospecifically from outer 1 and 3 positions of acylglycerols, such lipases hydrolyze triacylglycerol to give free fatty acids, 1,2-diacylglycerol, 2, 3 and 2-monoacylglycerols, example *Aspergillus niger* (Lawson *et al.*, 1994). Asahara (1993) worked on lipase from *Geotrichum sp.* FO401B and reported that it catalyzes the hydrolysis of triacylglycerides and release fatty acids selectively from the central position of acylglycerols.

Steriospecificity is the ability of lipase to discriminate between the enantiomers of a racemic pair. Steriospecificity is difficult to achieve by chemical methods (IUPAC Commission). Lipases have been used to perform stereospecific reactions to yield optically pure aliphatic and aromatic esters, alcohols, acids and lactones (Welsh *et al.* 2002). These reactions may be important as one isomer of certain compound may have desirable attributes than the other. For example the (R)-isomer of aspartame has a sweet taste while the (S)-isomer has a bitter taste attribute.

Substrates for lipases

Lipase catalyze various reactions since it has ability to act on wide range of substrates they may be artificial and natural, Table 2 summaries commonly used substrate for lipase assays.

Stability in organic solvents

Stability in organic solvents is desirable in synthesis reaction. From the available information it is concluded that lipases are generally stable in organic solvents with few exceptions of stimulation or inhibition (Gupta & Rath, 2004). Eventually high stimulation is noted in the presence of acetone, isopropanol and ethanol but was unaffected by methanol (Sharma *et al.*, 2009). Stability of lipases in different solvents is described in Table 3. Lescic *et al.* (2001) and Karadzic *et al.* (2006) worked



Table 2. list of commonly used substrate for lipase assays

Organism	Substrate	Reference
<i>Pseudomonas aeruginosa</i>	Soybean oil	Shabtai <i>et al.</i> (1992)
<i>P. aeruginosa</i>	butyl butyrate	Gaur <i>et al.</i> (2008)
<i>Pseudomonas sp.</i>	4-nitrophenyl myristate	Amada <i>et al.</i> (2000)
<i>Pseudomonas sp.</i>	Castor, coconut, corn, groundnut, linseed, neem, soybean, sunflower oils	Rathi, <i>et al.</i> (2000)
<i>Candida rugosa</i>	soybean oil	Sharma <i>et al.</i> (2001)
<i>Candida rugosa</i>	Triacetin	Pernas <i>et al.</i> (2001)
<i>Candida rugosa</i>	4-nitrophenyl decanoate	Chang <i>et al.</i> (2006)
<i>Candida rugosa</i>	4-nitrophenyl laurate	Chang <i>et al.</i> (2006)
<i>Penicillium candidum</i>	4-nitrophenyl palmitate & tributyrin	Ruiz <i>et al.</i> (2001)
<i>P. aurantiogriseum</i>	4-nitrophenyl caprate	Lima <i>et al.</i> (2004)
<i>Streptomyces rimosus</i>	Tween 20 and tween 60	Lescic <i>et al.</i> (2001)
<i>Aspergillus niger</i>	Tributyrin	van Heerden <i>et al.</i> (2002)
<i>A. carneus</i>	4-nitrophenyl myristate	Saxena <i>et al.</i> (2003)
<i>Bacillus megaterium</i>	4-nitrophenyl butyrate	Ruiz, <i>et al.</i> (2002)
<i>B. coagulans</i>	4-nitrophenyl palmitate	Kumar <i>et al.</i> (2005)
<i>B. coagulans</i>	4-nitrophenyl formate	Kumar <i>et al.</i> (2005)
<i>Bacillus sp.</i>	4-nitrophenyl caprylate	Sunna <i>et al.</i> (2002)
<i>Burkholderia multivorans</i>	4-nitrophenyl palmitate	Gupta <i>et al.</i> (2005)
<i>Psychrobacter sp.</i>	olive oil	Zhang <i>et al.</i> (2007)
<i>Aureobasidium pullulans</i>	peanut oil	Liu <i>et al.</i> (2008)

Table 3. Effect of solvents on lipases

Microorganism	Solvents	Residual activity
<i>Streptomyces rimosus</i> *, (Lescic <i>et al.</i> , 2001)	1,4-dioxane	100% stable
	Acetone	80%
	Acetonitrile	92%
	Dimethyl sulfoxide	63%
	Ethanol	91%
	N,N-dimethyl formamide	41%
	Tertrahydrofuran	No activity
<i>Pseudomonas aeruginosa</i> **, (Karadzic <i>et al.</i> , 2006)	Acetone	95%
	Butanol	10%
	Chloroform	130%
	Dimethylformamide	120%
	Ethanol	90%
	Hexane	100%
	Isopropanol	65%
Methanol	80%	

*18 h at 50% v/v; ** at 25% v/v, temperature 30°C

independently on effect of various solvents, time, temperature and concentration on *Streptomyces rimosus*, and *Pseudomonas aeruginosa* respectively. Residual activity ranges from 41% to 100% in *Streptomyces rimosus*, whereas, it was widely ranged from 10% to 100% and enhancement found in Dimethylformamide (120%) and Chloroform (130%) in *P. aeruginosa*. The effect of various solvents on lipases is given in Table 3.

Cellular localization of lipase

Mostly the prokaryotes secrete lipase extracellularly, however, eukaryotes synthesis cytosolic origin. Neugnot *et al.* (2001) reported cell bound lipases in *Candida parapsilosis* CBS 604. The cellular localization lipases in different organisms are given in Table 4.

pH and temperature kinetics

Lipases are active over broad pH and temperature range and they have molecular weight ranging from 94 to 840 kDa. From available literature it can be interpreted that generally lipases have neutral pH optima but the pH and temperature optima of lipases depends on the habitat of its sources. Lipases possess stability over a wide range from pH 4 to 11 and temperature optima in the range from 10 to 96°C. All organisms contain enzymes but they do not usually catalyze their substrate spontaneously, although certain physical treatments or conditions have been shown to activate the enzyme for example milk contain lipase but it unable to lipolysis milk fat spontaneously, the enzyme can be activated by prevailing some physical and chemical treatments such as heat, light, irradiation or chemicals impair the lipase activity (Chandan & Shahani, 1964).

Effect of metal ions

The activity of lipase may be inhibited or stimulated by cofactors. Divalent cations such as calcium often stimulated enzyme activity due to the formation of calcium salt of long chain fatty acids (Macrae & Hammond, 1985). Calcium stimulated lipases have been reported in the case of *Acinetobacter sp.* RAG-1 (Snellman *et al.*, 2002). In contrast, the lipase from *P. aeruginosa* 10145 (Finkelstein *et al.*, 1970) is inhibited by the presence of calcium ions. Further lipases activity is inhibited drastically by heavy metals like Ca⁺², Ni⁺², Hg⁺² and Sn⁺² and slightly inhibited by Zn⁺² and Mg⁺² (Patkar & Bjorkling, 1994). *P. aeruginosa* KKA-5 lipase hydrolyze castor oil in the presence of various metal chlorides, CaCl₂, AlCl₃ (group IIIB), CrCl₃ (group VIA) and MgCl₂ (group IIA) displays enhanced hydrolysis capability. When Cr⁺³ were used, hydrolysis of castor oil was four times faster than that of calcium, and 1.6 times faster with regards to Al⁺³. The chlorides of group VIII and alkali metals had no effect on hydrolysis (Sharon *et al.* 1998). The presence of chloride salts of Mg⁺², Cu⁺², Ca⁺², Hg⁺² and Fe⁺² resulted in a profound increase in the hydrolytic activity of the purified lipase. Interestingly, Hg⁺² ions resulted in a maximal increase in lipase activity but Co⁺² ions showed an antagonistic effect. The EDTA at a concentration of 150 mM markedly inhibited the activity of lipase. However, reconstitution of EDTA-quenched lipase with Hg⁺², Mn⁺² or NH⁺⁴ ions resulted in the restoration of the enzyme activity (Ghazi *et al.*) Lipase activity was enhanced in the presence of K⁺, Ca⁺² and Mg⁺² ions, but inhibited by Hg⁺² ions (Sharma *et al.*, 2009). The addition of Mg⁺² did not significantly stimulate lipase production. While many other metal ions including Ca⁺², Mn⁺², Ba⁺²,



Table 4 Cellular Localization of lipase

Organisms	Localization	Reference
<i>Homo sapiens</i>	Cyt	Waterman <i>et al.</i> (1998)
<i>Rattus norvegicus</i>	Cyt	Groener <i>et al.</i> (1981)
<i>Sus scrofa</i>	Cyt	Guibe-Jampel <i>et al.</i> (1987)
<i>Homo sapiens, Rattus norvegicus, Sus scrofa and Mus musculus</i>	ER	Dolinsky <i>et al.</i> (2004)
<i>Aspergillus niger</i>	Ext	Mhetras <i>et al.</i> (2008)
<i>Bacillus megaterium</i>	Ext	Ruiz <i>et al.</i> (2002)
<i>Penicillium candidum</i>	Ext	Ruiz <i>et al.</i> (2001)
<i>Pseudomonas aeruginosa</i> YS-7	Ext	Shabtai <i>et al.</i> (1992)
<i>Psychrobacter sp.</i> 7195	Ext	Zhang <i>et al.</i> (2007)
<i>Streptomyces rimosus</i>	Ext	Lescic <i>et al.</i> (2001)
<i>Rhizopus oryzae</i>	Ext	Sharma <i>et al.</i> (2001)
<i>Yarrowia lipolytica</i>	Ext	Yu <i>et al.</i> (2007)
<i>Pseudomonas fluorescens</i>	Int	Duong <i>et al.</i> (1994)
<i>Pseudomonas aeruginosa</i>	Int	Wohlfarth <i>et al.</i> (1988)
<i>Candida parapsilosis</i> CBS 604	CB	Neugnot <i>et al.</i> (2001)

Cyt= Cytosol, Ext= Extracellular, Int= Intraellular, CB= Cell bound, ER= endoplasmic reticulum.

Zn²⁺, metal ions, including Ca²⁺, Mn²⁺, Ba²⁺, Zn²⁺, Fe²⁺, and Cu²⁺ exerted inhibitory effects. However, lipase production was decreased slightly, to approximately 5%, with the addition of K⁺ and 30% decrease was observed in lipase production by S5 in an absence of potassium ions. The absence of magnesium ions (Mg²⁺) in the basal medium was also shown to stimulate lipase production. An alkaline earth metal ion, Na⁺, was found to stimulate the production of S5 lipase (Raja *et al.* 2006). The lipase activity in presence of a metal ions was compared with control including no metal ion whose activity was taken as 100% and the relative activities at 1mM of Cu²⁺, Hg²⁺, Pb²⁺, Co²⁺, Cd²⁺ and Li⁺ were 0.44, 24.4, 36.2, 49.1, 64.2, 90.0 and 98.2% respectively. Strong inhibition was observed with heavy metals such as Cu²⁺, Hg²⁺, Pb²⁺, Co²⁺ and Cd²⁺ in the *Todarodes pacificus* (Park *et al.*, 2007).

Table 5. List of industrial applications of lipases

Industry	Action	Product of application
Dairy food	Hydrolysis of milk, fat, cheese ripening, modification of butter fat	Development of flavoring agent in milk cheese and butter
Bakery food	Flavor improvement	Shelf life propagation
Beverages	Improved aroma	Alcoholic beverages e.g, sake wine
Food dressings	Quality improvement	Maysoine dressing and whippings
Health food	Transesterification	Health food
Meat and fish	Flavor development	Meat and fish product fat removal
Fats and oils	Transesterification and hydrolysis	Cocoa butter, margarine fatty acids, glycerol mono and diglycerides
Laundry	Reducing biodegradable strains	Cleaning cloths
Cosmetics	Esterification	Skin and sun-tan creams, bath oil etc
Surfactants	Replaces phospholipases in the production of lysophospholipids	Polyglycerol and carbohydrates fatty acid esters used as industrial detergents and as emulsifiers in food formulation such as sauces & ice creams.
Agrochemicals	Esterification	Herbicides such as phenoxypropionate
Pharmaceuticals	Hydrolysis of expoyester alcohols	Produce various intermediates used in manufacture of medicine.
Fuel industries	Transesterification	Biodiesel production
Pollution control	Hydrolysis and transesterification of oils and grease	To remove hard stains, and hydrolyze oil and greases.

Effect of light and irradiation

A very limited amount of work has been done to study the effect of light and irradiation upon lipase, especially

on the milk lipases but there are no reports on microbial, animal or plant lipases. Exposure of milk to bright sunlight, diffused daylight, mercury vapor lamp or ultraviolet light reduces the lipases activity 40 to 80% depending upon the length of exposure (Kay, 1946; Kannan, 1951). While Standhouders and Mulder (1958) observed that milk lipases was most sensitive to the blue rays of visible light. Tsugo and Hayashi (1962) reported that the lipase activity of milk was reduced by 70% at the irradiation dose of 6.6 X 10⁴r, Sahani and Chandan (1964) observed that only 1 - 38% activity of the purified enzyme was reduced when irradiated by as high as two and four megarads, respectively.

Lipase assay: Many procedures are available for assaying or characterizing hydrolytic activity of lipases (Chandan & Shahani (1964) and Pinsirodom & Parkin (2001)). Consequently, most lipase assays have been developed on the basis of measuring liberated fatty acids either specifically or nonspecifically. Alternatively the use of chromogenic or fluorogenic model substrates affords the option to spectrophotometry to directly and continuously follow the course of lipase reactions.

The titrimetric method is most common method for nonspecific measurement of fatty acids. In this method, native substrates i.e., triacylglycerols are hydrolyzed to yield fatty acids. Subsamples are withdrawn from reactive mixture at predetermined intervals, and reactivity is quenched by the addition of ethanol. The amount of fatty acids released during the reaction is determined by direct titration with NaOH to a thymolphthalein end point. The

quantity of fatty acid released in unit time is measured by the quantity of NaOH required to neutralize the mixture. The milliequivalent of alkali consumed is taken as a



measure of the activity of enzyme; this method is called as pH stat method (Chandan & Shahani, 1964).

Colorimetric method conceptually assays for lipase activity using the colorimetric method are similar to titrimetry in that liberated fatty acids are being measured; however, the colorimetric method is more specific for fatty acids. It involves the uses of chromogenic substrates, α -naphthyl acetate, for example, which upon hydrolysis yield or can be readily converted into a colored product, for colorimetric estimation. The color intensity is directly proportional to the lipase activity.

Spectrophotometric method in this method clearing of a fat emulsion as a function or the lipase concentration is measured spectrophotometrically but nowadays artificial or synthetic substrates are used such as p-nitrophenyl acyl substrates. The basis of this procedure is that lipases possess general esterolytic activity towards a variety of native and non-native carboxyl ester substrates. This method quantifies the level of p-nitrophenol released following the hydrolysis of p-nitrophenyl laurate substrate by lipase. Activity of lipase can be calculated by comparing samples A_{410} values to those of the standard curve prepared with p-nitrophenol (Pinsirodom & Parkin 2001).

Another method used extensively involves the quantitative extraction of free fatty acids from lipolyzed substrate either chromatographically or by extraction with chloroform, either or other organic solvents and titration with an alkali (Chandan & Shahani, 1964). Qualitative detection for lipolytic activity can be determined by using agar plates containing substrates for lipases with or without an indicator the most widely used methods are tributyrin agar plate, calcium-triolein agar plate and Rhodamine B agar plate. The first method uses tributyrin, a four carbon chain triglyceride, homogeneously emulsified in liquid agar media before pouring. Lipase producing microorganisms can be identified by a formation of clear zones around their colonies due to hydrolysis of tributyrin by the enzyme. However, the disadvantages of this method are that esterases can also hydrolyze tributyrin and give the same result. Calcium-triolein medium containing olive oil and calcium chloride can be used more specifically to detect for lipase activity. Lipase producing microorganisms are identified by a formation of white pellet of calcium oleate around the oil droplets in areas where colonies grow. Nonetheless, this method is sometimes difficult to detect for the microorganisms which produce low amount of lipases. The more sensitive method that uses medium containing rhodamine B as an indicator uses olive oil as a substrate for lipases reaction. Lipases activity can be visualized from fluorescence of the product; rhodamine B forms a fluorescent complex with free fatty acids, under UV light at 350 nm (Thongekkaew, 2006).

Recommendation and future prospects

Lipase has wide range of industrial application (Table 5). Lipase is one of the versatile groups of biocatalyst and

carries out novel reaction in both aqueous and non aqueous media. Lipase would be isolated from various sources and they have wide range of stability in varying conditions. Lipase has high range of specificity towards many substrates; some lipases are non specific while some are specific since they carry out wide range of reactions such as chemo, regio and enantio-selective transformation. Lipase has high potential regarding their catalytic behavior. Lipases have vast applications in various fields such as food, dairy detergent pharmaceutical, and agrochemical and oleochemical industries. It is a tool of choice for researchers but still it shares only about 3% of total market of all enzymes in world, so it is necessary to widen the usage of lipases. There is an urgent need to understand the mechanisms of action and still there is a lot of scope to search for new sources of lipases. On the basis of the present data it is reported here 38 bacteria, 14 fungi, 2 algae, 10 plants and 12 animals' lipases.

Concluding remarks

Algae are found to be major sources of lipids and fats and many lipase inhibitors are isolated from many fresh and marine water algae (Bitou *et al.*, 1999), so hypothetically it may contain lipases. *Chlorella pyrenoidosa* is found to be one of the algae for lipase (Wolfersberger & Pieringer, 1974). Considering the enormous data on bacterial, plant and animal origin lipase, based on their characterization, the algal group is relatively neglected. Therefore, more emphasis has to be given for characterization of algal lipases and hence further work is needed in these aspects.

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